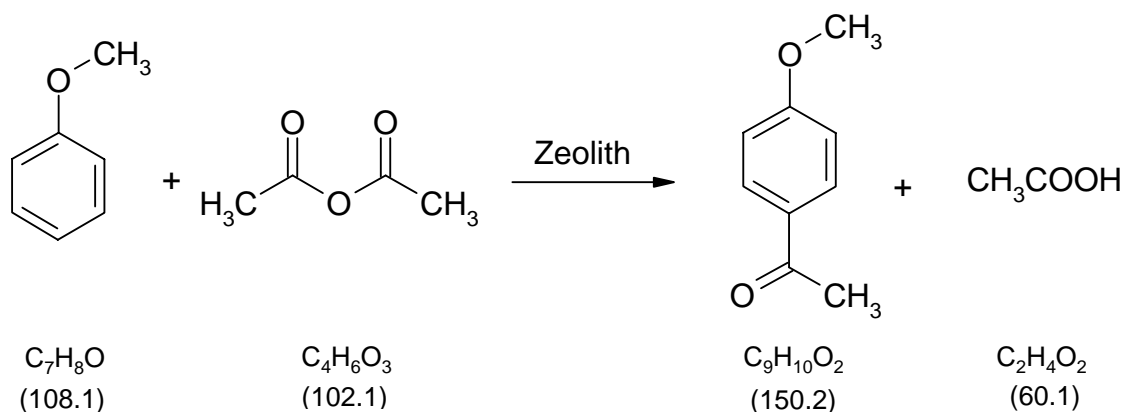


## 4010 Synthesis of *p*-methoxyacetophenone from anisole



### Classification

#### Reaction types and substance classes

electrophilic substitution of aromatics, Friedel-Crafts acylation, reaction of the carbonyl group in carboxylic acid derivatives

aromatics, carboxylic acid anhydride, acid catalyst

#### Work methods

heating under reflux, stirring with magnetic stir bar, filtering, evaporating with rotary evaporator, distilling under reduced pressure, heating with oil bath

### Instruction (batch scale 100 mmol)

#### Equipment

50 mL round-bottom flask, reflux condenser, Buechner funnel ( $\varnothing = 6.0$  cm), suction flask, heatable magnetic stirrer with magnetic stir bar, rotary evaporator, distillation apparatus, oil bath

#### Substances

anisole (bp 156 °C)	10.8 g (10.9 mL, 100 mmol)
acetic anhydride (bp. 140 °C)	15.3 g (14.2 mL, 150 mmol)
Zeolith H-BEA 25 (Süd-Chemie)	2.88 g
ethanol (bp 78 °C)	20 mL

#### Reaction

10.8 g (10.9 mL, 100 mmol) anisole, 15.3 g (14.2 mL, 150 mmol) acetic anhydride and 2.88 g Zeolith H-BEA 25 are placed into a 50 mL round-bottom flask containing a magnetic stir bar and fitted with a reflux condenser. The reaction mixture is heated with stirring for 6 hours to 150 °C.

**Work up**

The catalyst is filtered off using a Buechner funnel ( $\varnothing = 6.0$  cm) and washed with 20 mL ethanol. The filtrate is concentrated on a rotary evaporator.

Crude yield: 16.2 g; GC-purity 78% (see analytics)

The crude product is fractionally distilled under vacuum at 12 hPa (oil bath temperature up to 165 °C). Yield:

Fraction 1: bp 45 °C (12 hPa) (educt)

Fraction 2: bp 138 °C (12 hPa) (product); 11.6 g (77.2 mmol, 77%), white solid, mp 36 °C; GC-purity > 99%

**Comments**

When an equimolar amount of anisole and acetic anhydride is used, a reaction time of at least 20 hours is needed. By using an excess of 1.5 equivalents acetic anhydride, the reaction time is reduced to 6 hours.

**Waste management****Waste disposal**

Waste	Disposal
fraction 1 from distillation	organic solvents, halogen free
distillation residues	dissolve in a small amount of acetone, then: organic solvents, halogen free
ethanol used for washing	organic solvents, halogen free
residues from catalyst	solid waste, free from mercury

**Time**

6 hours for the reaction, 3 hours for the distillation.

**Break**

After heating under reflux and after concentrating with the rotary evaporator.

**Degree of difficulty**

Easy

**Instruction (batch scale 10 mmol)****Equipment**

10 mL round-bottom flask, reflux condenser, Buechner funnel ( $\varnothing = 2.0$  cm), suction flask, heatable magnetic stirrer with magnetic stir bar, rotary evaporator, kugelrohr distillation apparatus or microdistillatin apparatus, oil bath

**Substances**

anisole (bp 156 °C)	1.08 g (1.09 mL, 10.0 mmol)
acetic anhydride (bp. 140 °C)	1.53 g (1.42 mL, 15.0 mmol)
Zeolith H-BEA 25 (Süd-Chemie)	0.29 g
ethanol (bp 78 °C)	15 mL

**Reaction**

1.08 g (1.09 mL, 10.0 mmol) anisole, 1.53 g (1.42 mL, 15.0 mmol) acetic anhydride and 0.288 g Zeolith H-BEA 25 are placed into a 10 mL round-bottom flask containing a magnetic stir bar and fitted with a reflux condenser. The reaction mixture is heated with stirring for 6 hours to 150 °C.

**Work up**

The catalyst is filtered off using a Buechner funnel ( $\varnothing = 2.0$  cm) and washed with 15 mL ethanol. The filtrate is concentrated on a rotary evaporator.

Crude yield: 1.35 g

The crude product is distilled in a kugelrohr distillation apparatus at 25 hPa (oil bath temperature up to 165 °C).

Yield: 1.11 g (7.39 mmol, 74%), white solid, mp 35.6-37.5 °C

**Comments**

When an equimolar amount of anisole and acetic anhydride is used, a reaction time of at least 20 hours is needed. By using an excess of 1.5 equivalents acetic anhydride, the reaction time is reduced to 6 hours.

**Waste management****Waste disposal**

Waste	Disposal
distillation residues	dissolve in a small amount of acetone, then: organic solvents, halogen free
ethanol used for washing	organic solvents, halogen free
residues from catalyst	solid waste, free from mercury

**Time**

6 hours for the reaction, 1.5 hours for the distillation.

**Break**

After heating under reflux and after concentrating with the rotary evaporator.

**Degree of difficulty**

Easy

**Analytics****Reaction monitoring by GC**

Sample preparation:

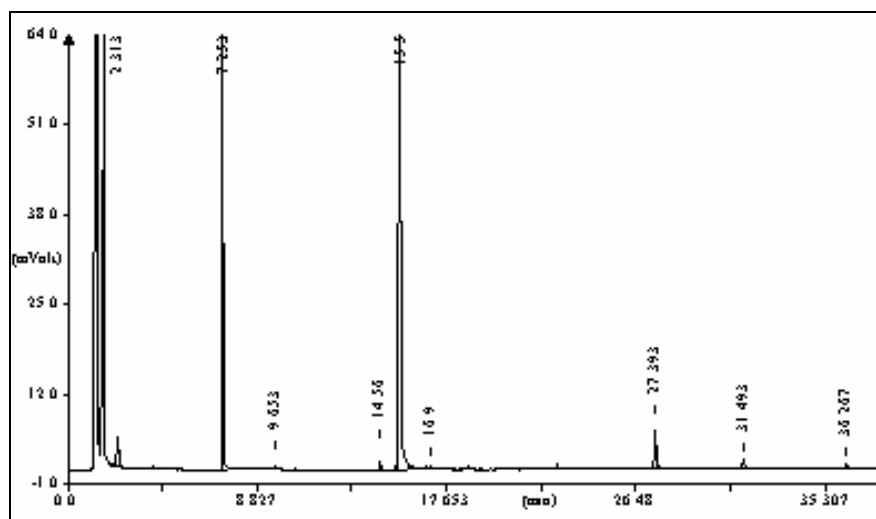
Using a Pasteur pipette, one drop of the reaction mixture is taken, diluted with 10 mL dichloromethane and filtered. From the filtrate, 0.2  $\mu$ L are injected.

GC conditions:

column: DB-1, L=28 m, d=0.32 mm, film=0.25  $\mu$ m  
 inlet: On-column injection, injected volume 0.2  $\mu$ L  
 carrier gas: H<sub>2</sub> (40 cm/s)  
 oven: 40 °C (5 min), 10 °C/min 240 °C (30 min)  
 detector: FID, 270 °C

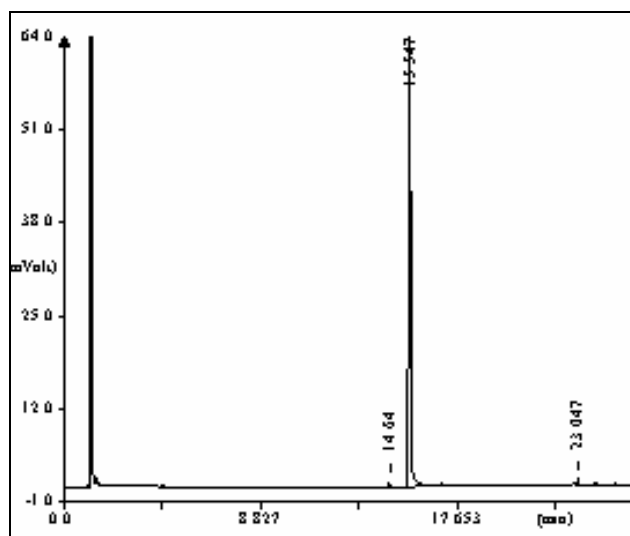
Percent concentration was calculated from peak areas

### GC of the crude product

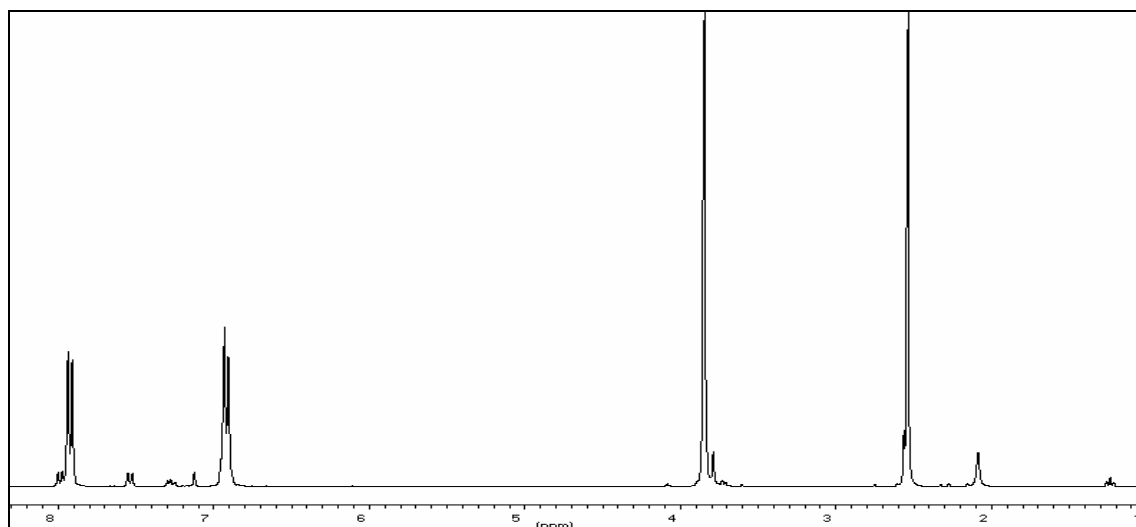
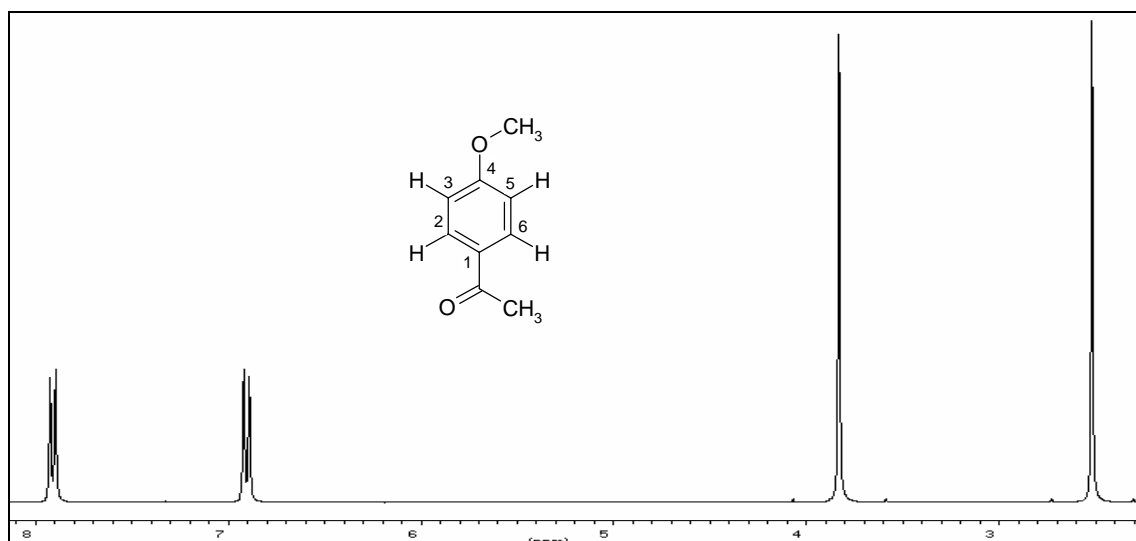


Retention time (min)	Substance	peak area %
2.31	educt (acetic anhydride)	5.8
7.25	educt (anisole)	14.4
15.50	product ( <i>p</i> -methoxyacetophenone)	77.8
27.39	side product	0.96

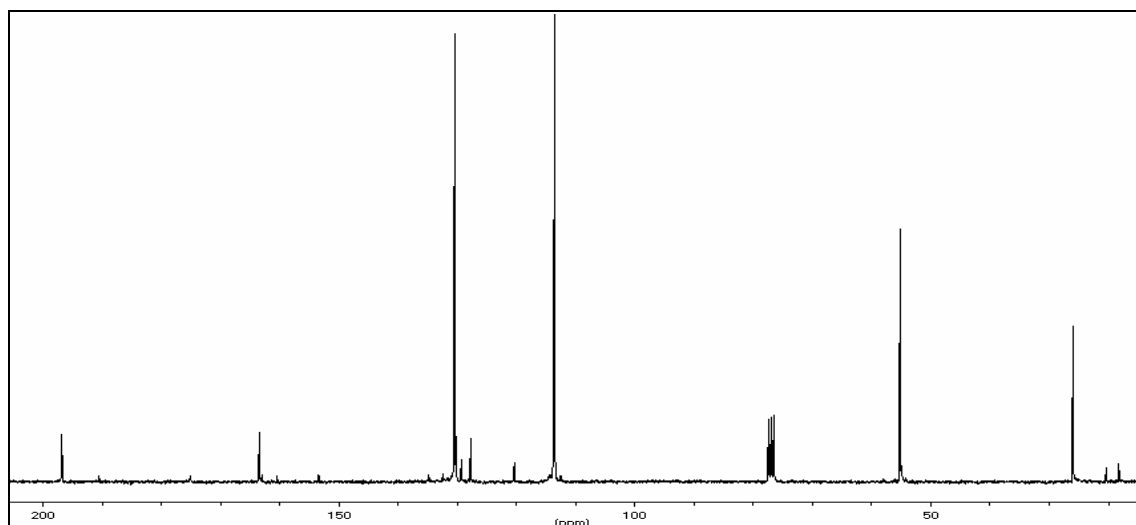
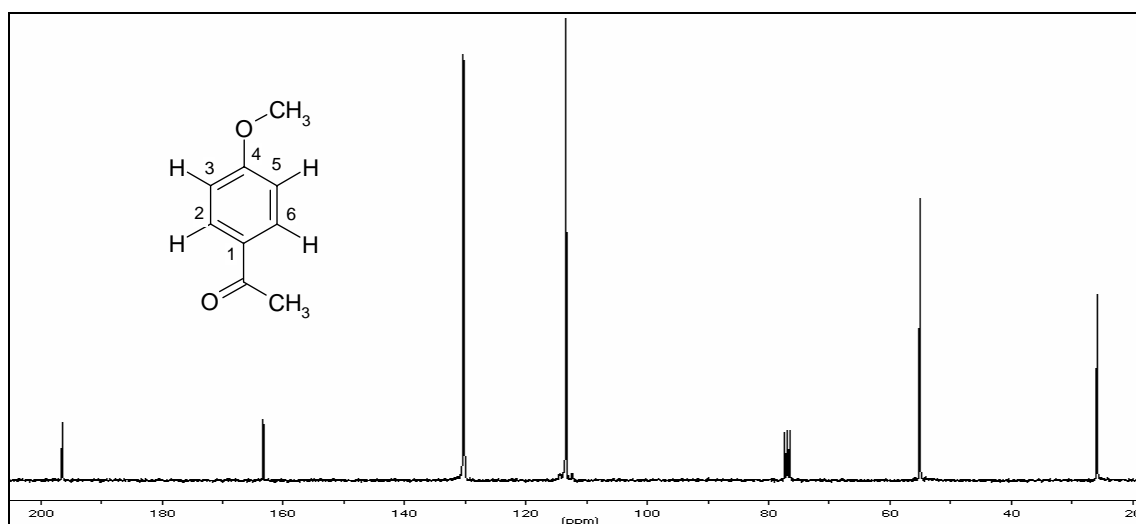
### GC of the pure product



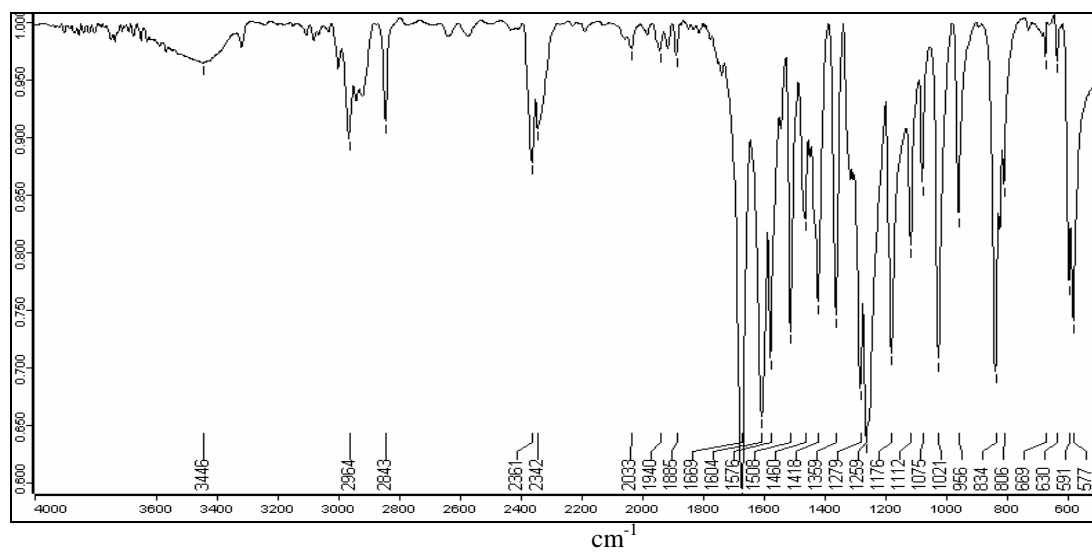
Retention time (min)	Substance	peak area %
15.55	product ( <i>p</i> -methoxyacetophenone)	99.7

**<sup>1</sup>H NMR spectrum of the crude product (300 MHz, CDCl<sub>3</sub>)****<sup>1</sup>H NMR spectrum of the pure product (300 MHz, CDCl<sub>3</sub>)**

$\delta$ (ppm)	Multiplicity	Number of H	Assignment
2.54	s	3	-CO-CH <sub>3</sub>
3.85	s	3	-O-CH <sub>3</sub>
6.91	m (AA')	2	3-H, 5-H
7.91	m (XX')	2	2-H, 6-H

**<sup>13</sup>C NMR spectrum of the crude product (75.5 MHz, CDCl<sub>3</sub>)****<sup>13</sup>C NMR spectrum of the pure product (75.5 MHz, CDCl<sub>3</sub>)**

$\delta$ (ppm)	Assignment
196.34	-CO-CH <sub>3</sub>
163.25	C-4
130.29	C-2, C-6
130.09	C-1
113.43	C-3, C-5
55.15	-O-CH <sub>3</sub>
25.98	-CO-CH <sub>3</sub>
76.5-77.57	solvent

**IR spectrum of the pure product (Film)**

( $\text{cm}^{-1}$ )	Assignment
3000	C-H-valence, arene
2964	C-H-valence, alkane
2843	C-H-valence, alkane, O-CH <sub>3</sub>
1617	C=O-valence, ketone
1604, 1576, 1508	C=C-valence, arene